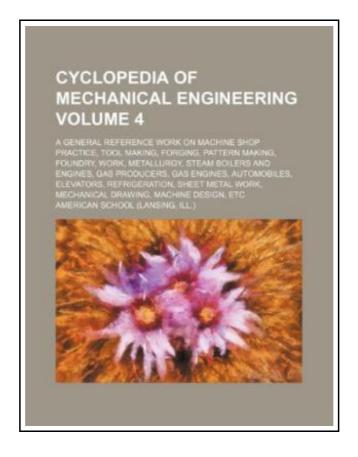
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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1908 edition. Excerpt: . in the moist steam = 11,076 + 35,952a; Therefore, 11,076 + 35,952x = 45,324 35,952x = 34,248 x = 0.952 That is, every pound of moist steam contains.952 lb. dry steam and.048 lb. moisture; or we may say there was 4.8 per cent of priming. The formula may be derived by the following algebraic work: Total heat in bbl. after condensation = W /; Total heat in bbl. before condensation = w qt; Total heat brought over by steam = W q, --w ql; Heat of liquid in condensed steam = (IV. w) q; Latent heat in dry steam = x (W--w) r; Total heat in moist steam = x (W--iv) r + (W--w) q. Therefore, x (IF--w) r + (IV--w) q = W q2--w g; xr (W--w) = W q2--w ql--W q + w q; or, transposing to a more convenient form, w(q-gj-lfr(?-?2) r (W-w) The use of this form of apparatus is not especially to be commended, for it is liable to error, and a slight discrepancy in the weights or the temperatures may cause a large error in the result. In the above calculations, no allowance is made for loss of heat through radiation. Separator Calorimeter. This instrument shown in Fig. 08, consists of a chamber A, into which is led a steam pipe D, bringing a sample of steam from the boiler or steam main. This pipe leads into an enlargement perforated with small holes, or into a chamber A...

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